# Peristome-Mimetic Curved Surface for Spontaneous and Directional Separation of Micro Water-in-Oil Drop



## **Abstract**

Inspired by ants that on the water covered peristome surface would helplessly slip on its surface. Here we report that instead of the "plug and go" separation model, tiny water-in-oil droplets can be separated into pure water and oil droplets through "go in opposite ways" on the curved peristome-mimetic surfaces in milliseconds without energy input. Furthermore, the spontaneously uni-directional transportation of the separated droplets is achieved on the curved peristome-mimetic surface.

## Methods

Guided by the natural designed structures of the peristome surface, here, we develop peristome mimetic curved surface by a 3D printing process and use it to separate tiny water-in-oil droplets. Digital light processing (DLP) three dimensional printing is used in our experiment to fabricate the inverse template of the peristome-mimetic surface. Without the need of the chemical modification or resin changing, the peristome-mimetic surfaces with low or high surface energies can be easily prepared through replication.

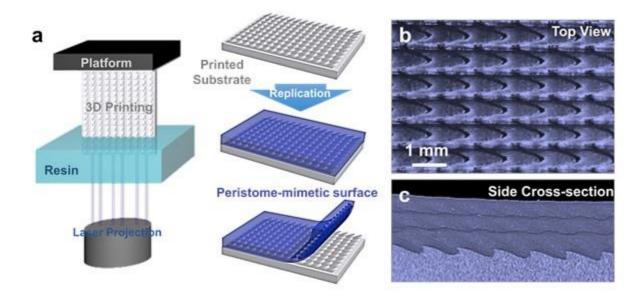


Fig. 1 The fabrication process. a, The inverse template is first fabricated by 3D printing. PDMS or PVA is then casted onto the printed substrate. Getting off the replica, the flexible peristome-mimetic film is finally prepared. b, Stereoscopic image of the peristome-mimetic surface, indicating that arch-shaped cavity structures arrayed into a pattern. c, Side cross-sectional views from the micro-CT observation.

# Liquid spreading behaviour

Peristome-mimetic PDMS surface shows hydrophobic with a water contact angle of 105° and superoleophilic with an hexadecane contact angle of less than 10°. Selective liquid transportation behavior is demonstrated by the high speed movie sequences when a water-in-hexadecane drop impacts on the peristome-mimetic PDMS surface. Even though the mass density of n-hexadecane, 0.77 g·cm-3, is lower than that of water, which is 1.03 g·cm-3, the oil would occupy the liquid-solid contact interface first, then uni-directionally spreads on the PDMS surface. In contrast to the directional wetting of the oil phase, water phase

transforms into a hemi-sphere shape and keeps its position at the rare side of the oil strip just below the nozzle.

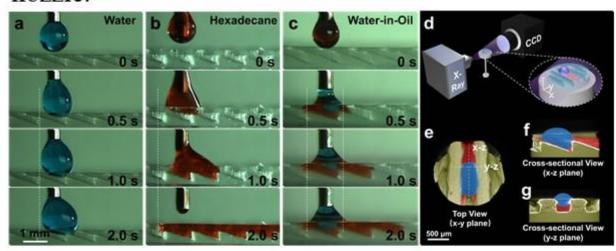


Fig. 2 The selective wetting behaviour of liquid on the peristome-mimetic surface. a-c, Time-lapse images of the water, hexadecane, water-in-hexadecane drops' spreading behaviour on the peristome-mimetic PDMS surface. d-f, 3D micro-observation of the selective liquid directional transportation on the peristome-mimetic surface.

#### Oil-water seperation

A novel oil-water separator based on double curved plates is devised to separate tiny oil-water mixture drops. we placed two curved peristome-mimetic surfaces with the structured sides face to face, oil would transport along the bottom surface in one direction and water spreads in another direction, just after the water-in-oil droplet contacted the surface. Controlling the parameters of the peristome-mimetic surfaces, our device can spontaneously separate, for the first time, several microliters sized water-in-oil droplets into controllable sized pure oil and pure water droplets with volumes ranging from nanolitres to microliters in controlled speeds even within milliseconds.

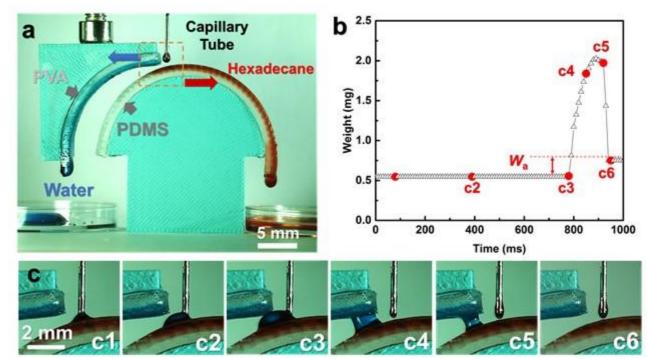


Fig. 3 Nano-liter scaled water-oil separation on the curved PDMS and PVA peristome-mimetic surface. a, Optical images depicting the experimental setup of the water-oil separation device. b, High sensitivity microbalance system recording the force during the separation process. c, Time lapse images showing the process of nano-liter water separated from oil coated PDMS surface and transporting on the PVA surface.

### **Publications**

- 1. Uni-Directional Transportation on Peristome-Mimetic Surfaces for Completely Wetting Liquids, Angew. Chem.-Int. Edit., 2016, 55(48), 14988-92.
- 2. Peristome-Mimetic Curved Surface for Spontaneous and Directional Separation of Micro Water-in-Oil Drop, Angew. Chem.-Int. Edit., 2017, early online.